# Outline Facts of Chemistry; With Exercises Intended Chiefly for Pupils in Government Science Classes



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## Reviews

Complete guideline for ebook enthusiasts. It really is loaded with knowledge and wisdom Once you begin to read the book, it is extremely difficult to leave it before concluding.

(Delilah Hansen)

# OUTLINE FACTS OF CHEMISTRY; WITH EXERCISES INTENDED CHIEFLY FOR PUPILS IN GOVERNMENT SCIENCE CLASSES



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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English. Brand New Book \*\*\*\*\*\* Print on Demand \*\*\*\*\*\*. This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1866 edition. Excerpt: .dissolving chalk in hydrochloric acid. It deliquesces rapidly in air, and is often used after fusion to remove watery vapour from gases; has a bitter taste. 287. Oxychloride of Calcium is prepared by boiling the chloride with quicklime, and filtering whilst hot. Its formula is CaCl2 3CuO. 15H3 0. 288. Chloride of Lime. This bleaching material is obtained by passing chlorine into chambers containing hydrate of lime in fine powder. See Bleaching, Par. 61. Its composition is supposed to be Ca Cl2 Ca 2ClO. 289. Fluoride of Calcium, or Fluorspar, Ca F2. Sp. Gr. 3.14. This occurs abundantly in the lead mines of Derbyshire, Cumberland, c., and is called blue John. It is most frequently found of a blue, green, or white colour; the blue varieties become red, and brown by heat; when gently heated, the mineral becomes phosphorescent. The artificial compound may be prepared by saturating hydrofluoric acid with carbonate of lime. 290. Sulphate of Calcium, Ca S04; Sp. Gr. 2.95. Native Sulphates: --Anhydrite, containing no water; the hydrates containing two atoms of water, as alabaster, gypsum and selenile. Obtained artificially, when a salt of calcium in solution is precipitated by sulphuric acid. It requires 400 parts of water to dissolve it. Heated to 300, it loses 20 per cent-of water, and forms plaster of Paris; plaster of Paris mixed with size and water, hardens, and hence is used for artificial marble, stucco, busts, figures, c.; sulphate of calcium is often found in spring...

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