



Chemical Primer; An Elementary Work for Use in High Schools, Academies, and Medical Colleges

By Simeon Pease Meads

Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1885 Excerpt: .orange, Sn S brownish-black, Sn S2 yellow. Hydrogen sulphide (H₂S sulphuretted hydrogen) is much used in the laboratory to precipitate metals, as sulphides. (See Analytical Charts.) H₂S is readily inflammable, as may be shown by igniting in test-tube. Note.--Hydrogen sulphide has a slight acid reaction and was called by the old chemists hydrosulphiiric acid. It unites with many of the bases to form sulphides, and these sulphides might be classed as binary salts. For reasons which need not be explained here, chemists do not no class sulphides in this way, but consider them as analogous to oxides. Carbon disulphide (C S₂), a volatile, colorless, inflanw mable liquid, may be produced by passing sulphur over red-hot coals. It is an excellent solvent, dissolving readily S, P, I, and many organjp substances. It refracts light powerfully, and hence is often used in filling prisms. The impure...



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