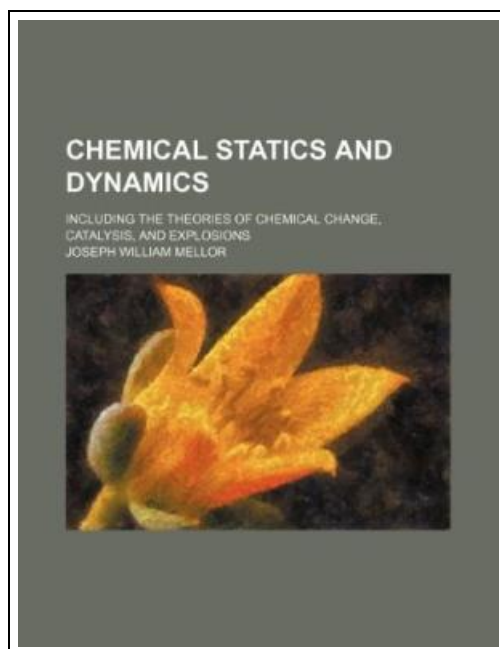


Chemical Statics and Dynamics; Including the Theories of Chemical Change, Catalysis, and Explosions



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Reviews

Simply no words and phrases to clarify. It really is full of knowledge and wisdom You wont feel monotony at at any moment of the time (that's what catalogs are for relating to when you question me).




(Paolo Spinka)

CHEMICAL STATICS AND DYNAMICS; INCLUDING THE THEORIES OF CHEMICAL CHANGE, CATALYSIS, AND EXPLOSIONS



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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1909 Excerpt: . attempt is made to measure the rate of combination of hydrogen and oxygen. Berthelot¹ found that the rate of combination of hydrogen and oxygen, at 250-300, was accelerated by the presence of barium hydrate, by alkalies, and by traces of manganese salts. Now, these substances may be regarded as decomposition products of glass, and are therefore always present on the surface of glass vessels, and Berthelot thinks that the discordant results obtained when the speed of the reaction is measured in different vessels might be due to the presence of varying amounts of the decomposition products of the glass. Reactions in liquid menstua are not so sensitive to the nature of the walls of the vessel as gaseous reactions. But still the nature of the vessel does exercise a measurable influence on the course of some reactions. For example, Rayman and Sulc² find that metal vessels exert a specific influence on the inversion of cane sugar. Polished platinum vessels, however, 1 M. Berthelot, *Compt. Rend.*, 125. 271, 1897. 1 B. Rayman and O. Sulc, *Zeit. phys. Chem.*, 21. 481, 1896; 33. 47, 1900; F. Plzak and B. Husek, *ib.*, 47. 733, 1904. do not influence the decomposition of hydrogen peroxide.¹ The rate of reduction of Fehling s solution by invert sugar is augmented by the use of vessels with a large internal capacity.² The cuprous oxide formed during the reaction also seems to accelerate the velocity of the reduction. 80. J. J. Thomson s Surface...

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