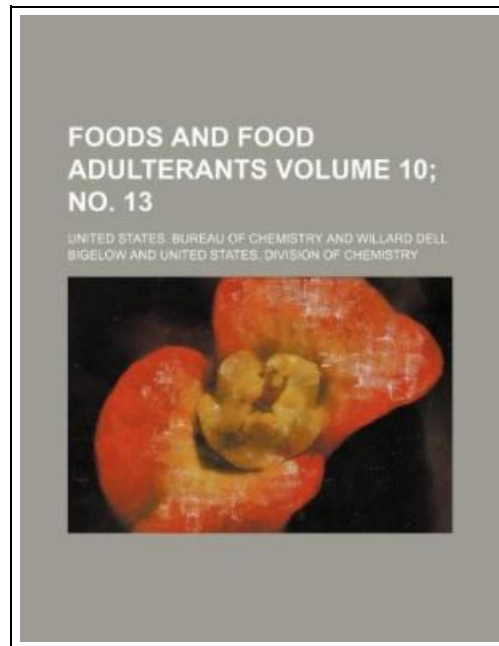


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Reviews

Totally one of the better publication I have actually read through. It really is rally fascinating throggh studying time period. Its been printed in an extremely simple way and is particularly just following i finished reading through this ebook in which basically modified me, modify the way i think.

(Mrs. Maudie Weimann)

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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1902 Excerpt: .acid. It is found convenient to treat a somewhat larger weight of meat and divide the filtrate into different portions for the two determinations. The filtrate is transferred to a short-necked Neviere and Hubert, Mon. sci., 1895 (4), 9, 324. The methods given for the detection of benzoic acid can not be employed in the presence of saccharin. flask and subjected to distillation. The first portions of the distillate are used in the detection of sulphurous acid as directed hereafter. The last portion is employed in the detection of benzoic acid according to one of the following methods: MOHLER S METHOD.b The filtrate described above is neutralized with sodium hydroxid, evaporated to dryness, treated with 2 or 3 cc of strong sulphuric acid, and heated until white fumes appear. By this means benzoic acid is converted into sulphobenzoic acid. A few crystals of potassium nitrate are added and the heating continued until the solution is almost or quite colorless. This causes the formation of meta-dinitro-benzoic acid. When cool the acid is diluted with water, ammonia added in excess, and the mixture transferred to a test tube. A drop or two of ammonium sulphid is now added, taking care that the liquids shall not mix. The nitro compound is converted into ammonium metadiamido-benzoate, which possesses a peculiar red color. This reaction takes place immediately, and is seen at the surface of the liquid without stirring. Salicylic acid will sometimes give the same reaction, but only after waiting...



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