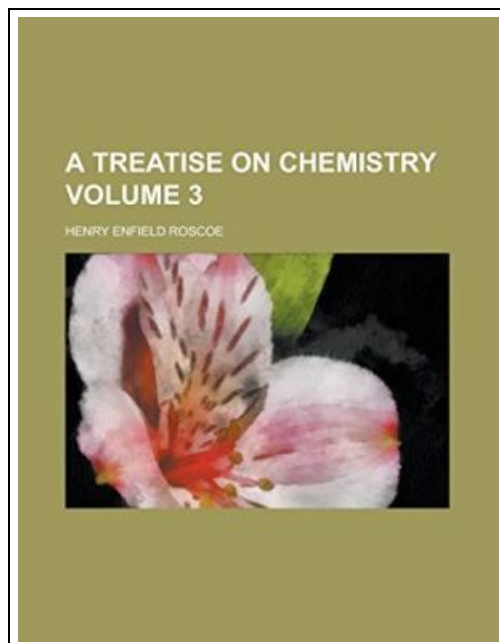


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RareBooksClub. Paperback. Book Condition: New. This item is printed on demand. Paperback. 144 pages. Dimensions: 9.7in. x 7.4in. x 0.3in. This historic book may have numerous typos and missing text. Purchasers can usually download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1890 edition. Excerpt: . . . Ann. Chem. Pharm. lxxxvii. 376. Aniline is chiefly used in the colour industry. In the laboratory it is frequently employed as a solvent, since it dissolves many substances--e. g. indigo-blue--which are insoluble in the ordinary reagents. SALTS OF ANILINE. 1027 Although its basic properties are feeble, aniline combines with acids, forming well-crystallizable salts. Hence Unverdorben, as has been already remarked, gave to it the name of crystalline. The salts of aniline have been carefully examined, but we shall here only refer to those which have received technical applications, or are specially characteristic. Aniline hydrochloride, $C_6H_7N.C_1H$, is readily soluble in water, and crystallizes in needles or large plates, which melt at 192, and sublime unchanged. It is prepared on the large scale, and generally called aniline salt; the platinum chloride, $(C_6H_7N.HC_1PtC_1)$ crystallizes from hot water in yellow needles. Aniline sulphate, $(C_6H_7N)_2S_0_4H_2$, forms a crystalline powder, readily soluble in water, slightly so in alcohol, and insoluble in ether. This last property is taken advantage of in the separation of aniline from methylaniline. Aniline nitrate, $C_6H_7N.N_0_3H$, crystallizes in large rhombic prisms or needles, which decompose at 190 with formation of nitraniline and other products (Bechamp). Aniline oxalate, $(C_6H_7N)_2C_2_0_4H_2$, is very slightly soluble in cold water, and crystallizes from a hot solution in fine triclinic prisms; it is slightly soluble in alcohol, and insoluble in ether. 1 fmn dbomme. loc. eit. Aniline phenate, $C_6H_6.NH_2.C_6H_6O$, is formed by heating a mixture of aniline and phenol...

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