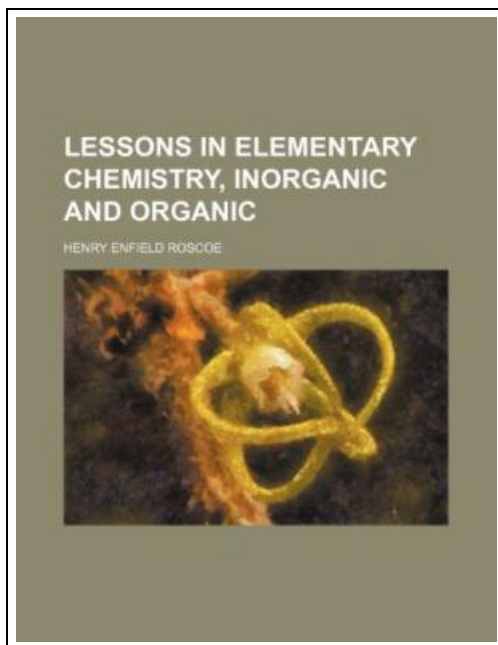


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1868 edition. Excerpt: .which a solution of cobalt chloride imparts to these salts when heated before the blowpipe. 3. Cadmium. Symbol Cd. Combining Weight 112. Specific Gravity 86. This is a comparatively rare metal, occurring in certain zinc ores. In its chemical relations it closely resembles zinc. It is, however, more volatile than the latter metal, and therefore distills over first in the preparation of zinc. Cadmium is a white ductile metal, melting at 3150; it may be easily distinguished and separated from zinc by yielding a bright yellow sulphide insoluble in hydrochloric acid. The metal takes fire when heated in the air, forming a brown oxide, Cd O. The chloride and sulphate are soluble well-crystallizing salts. Cadmium iodide is occasionally used in photography, and the yellow sulphide has been employed as a pigment. Class V. Iron Class.--1, Manganese. 2, Iron. 3, Cobalt. 4, Nickel. 5, Chromium. 6, Uranium. 7, Indium. 1. Manganese. Symbol Mn. Combining Weight 55. Specific Gravity 80. Manganese occurs in nature as an oxide, and it can be obtained, though with difficulty, in the metallic state by heating the oxide very strongly with charcoal. The metal is of a reddish-white colour; it is brittle, and hard enough to scratch glass. It decomposes water at the ordinary temperature with evolution of hydrogen; it cannot be preserved in the air without undergoing oxidation, and must be kept under naphtha, or in a sealed tube; it is slightly magnetic, and, like iron, combines with carbon and silicon. Metallic...

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